Question 1

1	1(d)	yttrium < nickel < samarium < strontium (2)	2
		if 2 marks not scored: 1 mark if one consecutive pair reversed OR strontium < samarium < nickel < yttrium	

Question 2

2(d)	niobium < nickel < iron < magnesium (2)	2	l
	1 mark if one pair reversed		l

Question 3

3(d)	lead <nickel<iron<magnesium (2)<="" th=""><th>2</th></nickel<iron<magnesium>	2
	if 2 marks not scored: 1 mark for 1 pair adjacent reversed OR magnesium <iron<nickel<lead< td=""><td></td></iron<nickel<lead<>	

Question 4

4(d)	mercury < tin < iron < magnesium (2)	2
	1 mark if one pair reversed	

Question 5

5(c)(i)	cobalt < zinc < magnesium < barium	2
	if 2 marks not scored, 1 mark for all reversed / one consecutive pair reversed	

Question 6

	6(c)	copper , zinc , magnesium , potassium (2)	2	
ı		if 2 marks not scored: 1 mark for one consecutive pair reversed		l

Question 7

7(e)	tin < nickel < magnesium < sodium (2)	2
	IF two marks not scored, 1 mark for one adjacent pair reversed.	

Question 8

8(e)	M1 aluminium oxide layer (1)	2
	M2 (oxide layer) is unreactive (1)	

Question 9

ſ	9(c)(i)	iron is more reactive than copper	1
	9(c)(ii)	$Fe + CuSO_4 \rightarrow Cu + FeSO_4$	1
	9(c)(iii)	electrolysis	1

Question 10

10(b)(i)	any metal above zinc in reactivity series	1
10(b)(ii)	any metal below iron in reactivity series	1